



Woudschoten werkgroep 2017

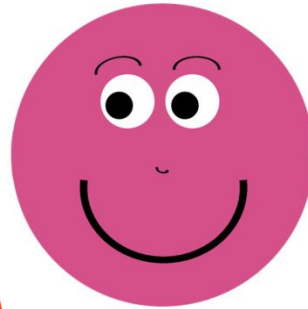
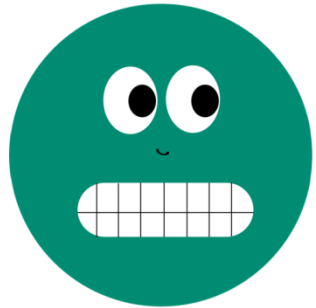
Quantumwereld

Wat kunnen we leren van andere landen?

Kirsten Stadermann



- Introductie van Quantumwereld in CE





KWANTUM

Zoeken in de hele catalogus

Zo

Home

Gordijnen &
 raamdecoratie

Vloer

Tuin

Verlichting

Behang &
 verf

Meubelen

✓ Altijd de laagste prijs

✓ 100% Tevredenheidsgarantie

✓

NERGENS GOEDKOPER



Wat leren ze in andere landen over quantumfysica?



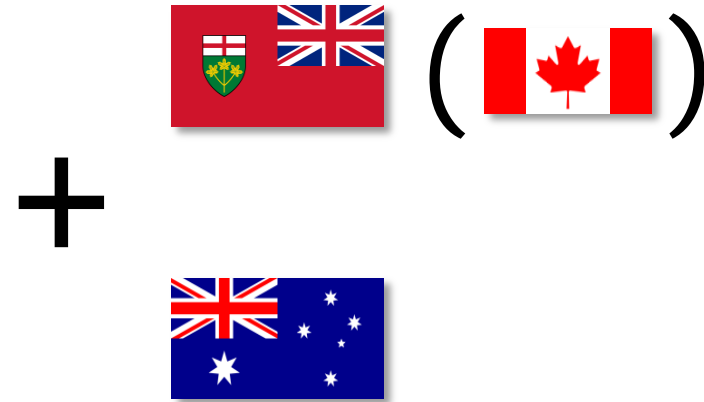
- Background
- Research question
- Method
- **Findings**
- Analysis
- Conclusion
- Future research
- Acknowledgements
- Questions / feedback

Quantum Physics in Secondary Schools		Norway	Finland	Sweden	Denmark	Netherlands	Belgium (Flemish community)	UK (England and Wales)	Ireland	Germany (Lower Saxony)	Germany (NRW)	Germany (Hesse)	Germany (Rhineland Palatinate)	Germany (Baden Württemberg)	Germany (Saxony)	Germany (Bavaria)	Austria	Italy (Liceo Scientifico)	France	Spain	Australia	Canada, (Ontario)	
Comparison of some intended learning outcomes in official curricula or syllabi of 15 different countries																							
29-06-2017																							
<input checked="" type="checkbox"/> = compulsory <input type="checkbox"/> = optional																							
General Inform.	Quantum Physics in the curriculum since	?	?	?	?	2014	?	?	-	<1960	<1960	<1960	<1960	<1960	?	<1960	?	?	?	?	?	?	
	Grade in which Quantum Physics is taught	12	12	12	12	11/12	11/12	11/12	-	11/12	11/12	11/12	11/12	11/12	11/12	10/11/12	11/12	11/12	11/12	11/12	11/12	11/12	
	Implementation year of investigated curriculum/syllabus (first exam)	2006	2005	2011	2008	2016	2015	2014	1999	2009	2014	2016	1998	2016	2012	2011	2014	2010	2012	2015	2016	2009	
	Name/abbreviation of physics course	1 2	1 B	2 3	A B	na	F K	A S	D O H	G E	G L	G L	G L	L 2 4	G L	Ph	Ph	FI	FI	F2	4U	4U	
	Written leaving exam, centrally set	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■
	Written leaving exam, set by teacher/school	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■
Oral examination (possible)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
Topics mentioned in the official curriculum of a course: (selected items)																							
Skills	S1 Use of models in physics	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	S2 Use of IT simulations	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	S3 Oral communication skills (discussion, presentation)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
NOS	N1 Scientific method (formulation, test and modification of hypotheses)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	N2 History of physics	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	N3 Reflection on development of models in physics (paradigm shift)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	N4 Physics in relation to social, ethical, philosophical or religious issues	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
Quantum Physics	Q1 Black body radiation	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q2 Bohr model of (hydrogen) atom (= electrons on certain allowed orbits)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q3 Discontinuity (energy levels in atom, emis./absorption, line spectrum)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q4 Interactions between light and matter (photoel. effect, Compton effect)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q5 Matter waves (de Broglie wavelength, interference of electrons)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q6 Wave-particle duality (complementarity, double slit experiment)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q7 Statistical predictions (probability, Copenhagen interpretation)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q8 Uncertainty (Heisenberg's Principle)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q9 Technical applications (e.g. semiconductor, LED, laser, SEM)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q10 Sum over paths approach (phasor representation)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q11 Schrödinger equation (one dimensional time independent)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q12 Detection probability = sq. of wave function amplitude or phasor length	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q13 One dimensional model (potential well, particle in a box)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q14 Tunneling, Alpha decay	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q15 Atomic orbital model (electron cloud, 3-dimensional potential well)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q16 Pauli exclusion principle (structure of Periodic Table)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
	Q17 Philosoph. consequences (interpretations, Schr.'s cat, observer role)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	
Q18 Entanglement (interconnectedness, non-locality, EPR-experiment)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■		
Q19 Modern research (e.g. quantum computing, quantum cryptography)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■		
Q20 Experimental determination of h (e.g. with LED)	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■	■		

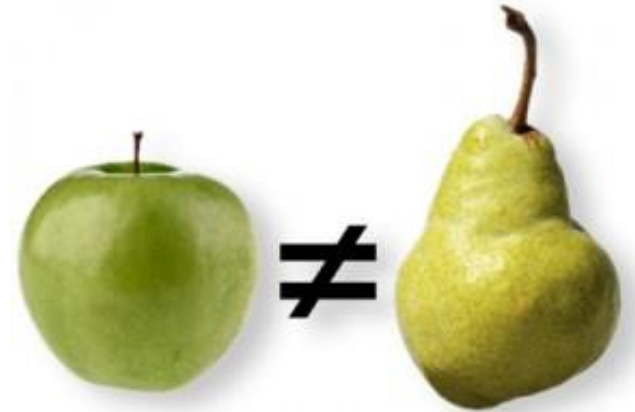


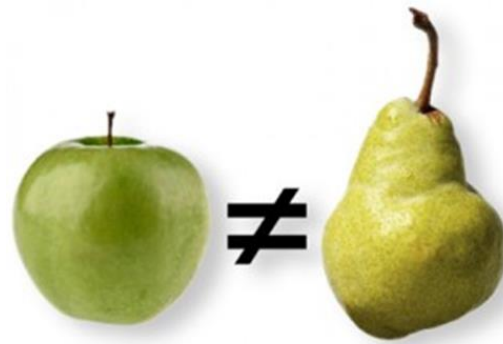
Countries of Analysed Curricula

- Background
- Research question
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Vergelijken





- In welke klas leren ze QF?
- Hoeveel tijd hebben leerlingen ervoor?
- Is het een keuzeonderwerp of verplicht?
- Hoe ziet het eindexamen er uit?



Quantum Physics in Secondary Schools

Comparison of some intended learning outcomes
in official curricula or syllabi of 15 different countries

29-06-2017

■ = compulsory
□ = optional

	Norway	Finland	Sweden	Denmark	Netherlands	Belgium (Flemish community)	UK (England and Wales)	Ireland	Germany (Lower Saxony)	Germany (NRW)	Germany (Hesse)	Germany (Rhineland Palatinate)	Germany (Baden Württemberg)	Germany (Saxony)	Germany (Bavaria)	Austria	Italy (Liceo Scientifico)	France	Spain	Australia	Canada, (Ontario)	
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Kernbegrippen: paars

‘Karaktertrekken van QF’

- Stochastisch gedrag en waarschijnlijkheid (Q7)
- Superposition en Interferentie (Q5, Q6)
- Resultaat en invloed van metingen (Q6, Q7, Q8)
- Complementariteit (Q5, Q8)

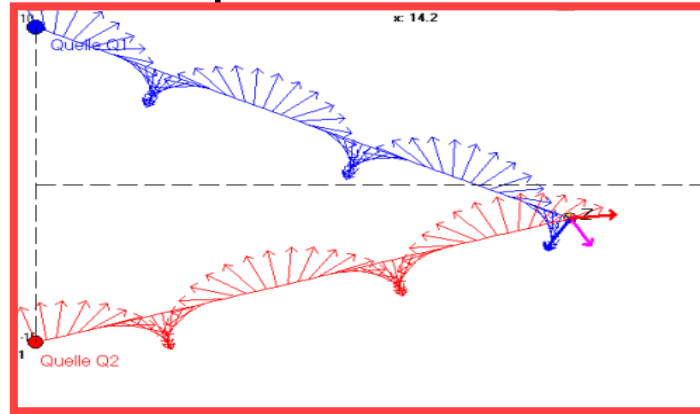
- Entanglement (Q18)

Wiskundige beschrijvingen?

- differentiaalvergelijkingen

$$-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi(x,t)}{\partial x^2} + V(x,t)\Psi(x,t) = i\hbar \frac{\partial \Psi(x,t)}{\partial t}$$

- of phasor representatie



Wiskundige beschrijvingen

- Alleen in Duitse deelstaten
- Lange traditie

	UK (England & Wales)	Ireland	Germany (Lower Saxony)	Germany (NRW)	Germany (Hesse)	Germany (Rhineland Palatinate)	Germany (Baden Württemberg)	Germany (Saxony)	Germany (Bavaria)	Austria
Year	<1960	<1960	<1960	<1960	<1960	<1960	<1960	?	<1960	?
Grade	11/12	11/12	11/12	11/12	11/12	11/12	11/12	11/12	10/11/12	11/12
Year	1999	2009	2014	2016	1998	2016	2012	2011	20*	
Code	H	G	E	G	L	G	L	G	L	Ph
Other										

**Al meer dan 50
jaar in het vwo-
curriculum**



Waarom quantenfysica?

- **Belangrijke tak van de moderne natuurkunde**

Als de quantenfysica in staking zou gaan, zou elk moderne apparaat uitvallen met alle gevolgen van dien.

Het is zo'n grote en belangrijke theorie als de evolutietheorie en elk mens met een goede schoolopleiding moet ervan weten.



Waarom quantenfysica?

- **Belangrijke tak van de moderne natuurkunde**
- **Motivatie**
Quantum is cool







Waarom quantenfysica?

- **Belangrijke tak van de moderne natuurkunde**
- **Motivatie**
- **Belangrijk voor het begrijpen van wetenschap**



Bungum (Norway)

“There have been discussions on the topic entangled photons, some teachers and physicists don’t like to have it in the curriculum since there are no clear answers to what the phenomenon is.

But this is why it was put in the curriculum, to show that physics has unsolved problems.”



Nature of Science (NoS) = de natuur van de wetenschap

Science is built up of facts, as a house is built of stones; but an accumulation of facts is no more a science than a heap of stones is a house.

Henri Poincare

“Een verzameling van wetenschappelijke feiten is net zo min wetenschap als een stapel stenen een huis is.”



Misvattingen over NoS:

- Er bestaat één wetenschappelijke methode.
- Wetenschappers volgen alleen die regels
- Wetenschappelijke kennis is eenduidig en onveranderlijk”
- OF: Omdat wetenschappelijke kennis kan veranderen, kun je het niet vertrouwen (“Het is ook maar een mening”)
- Wetenschappers zijn niet creatief.
- Wetenschap inspireert technologie, maar niet andersom.



REGENESIS TOUCH[®]

De ultieme Quantum-Healing methode

DE REGENESIS CONNECTIE NAAR ONZE GEZONDHEID

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...use QuantumVET!"*



In feite kan elke les natuurkunde
een les over NoS zijn.
Maar met quantenfysica zijn we
gedwongen om het ook te
behandelen.

Oostenrijk





Noorwegen:

- ReleQuant

Quantum Physics

Physics aims to describe how the world works, from the greatest galaxies down to the smallest particles. Many physicists would like one single theory that can describe all natural phenomena. We do not have such a theory today.

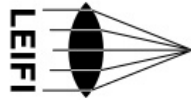
This learning resource was developed by the research and development team in [the project ReleQuant](#), based at the University of Oslo, the Norwegian Centre for Science Education and the Norwegian University of Science and Technology. It has received support from the Research Council of Norway and the Olav Thon Foundation.





ReleQuant's objectives

- to develop digital resources in quantum physics and general relativity that are in line with the Norwegian physics curriculum
- let pupils develop their understanding of physics collaboratively by „talking physics“ and „writing physics“
- make use of IT resources such as animations and simulations where appropriate
- motivate by showing the relevance of modern physics for pupils' interests and for their educational and career plans
- encourage reflections on historical and philosophical perspectives on science - for example by presenting physicists and their various interpretations of physical phenomena
- clarify the principles that classical physics is built upon to make clear that modern physics drastically changes these underlying principles



Aktuelle Suche

- quanten

Filtern nach Teilgebiet:

- Kern-/Teilchenphysik (13)
- Atomphysik (6)
- Quantenphysik (1)

Filtern nach Typ:

- Aufgaben (19)
- Downloads (1)
- Grundwissen (1)

Sortieren nach Relevanz

Suchergebnisse für: quanten



23 Treffer gefunden (0.414 Sekunden)



Atome - Wellen - Quanten

Teilgebiet: Quantenphysik Themenbereich: Quantenobjekt Photon Typ: Links

Atome - Wellen - Quanten



Was sind Quanten oder was sind sie nicht? (Interaktives Tafelbild)

Tafelbilder

Typ: Download

Quanten-, Wellen- und Teilchenphänomene am Mach-Zehnder-Interferometer Im Tafelbild werden Quanten als ein neues physikalisches Modell vorgestellt, welche sich deutlich vom Teilchen- und...



Duitsland: milq

Quantenphysik in der Schule

Die Quantenmechanik wird nicht nur von Schülern als schwierig empfunden. Sie ist es wirklich. Den außerordentlich schwierigen begrifflichen Problemen, die sie aufwirft, muss sich jeder Unterrichtende stellen. Das Internetportal **milq** richtet sich an alle, die in der Schule Quantenphysik unterrichten und/oder sich gerne noch etwas ausführlicher damit beschäftigen wollen.

Im Menü können Sie folgende Bereiche auswählen:

Der milq-Lehrgang: Hier findet man an den Unterricht angepasstes begriffliches Hintergrundwissen.

Der milq10-Lehrgang: Hier entsteht eine geänderte Version von milq für die 10 Jahrgangsstufe

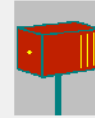
Das Schülerprogramm zur Quantenreflexion (S·P·Q·R) bietet ein auf dem milq-Lehrgang basierendes Unterrichtskonzept für die Oberstufe.



Der quantenmechanische Doppelspaltversuch

Datei Optionen Hilfe

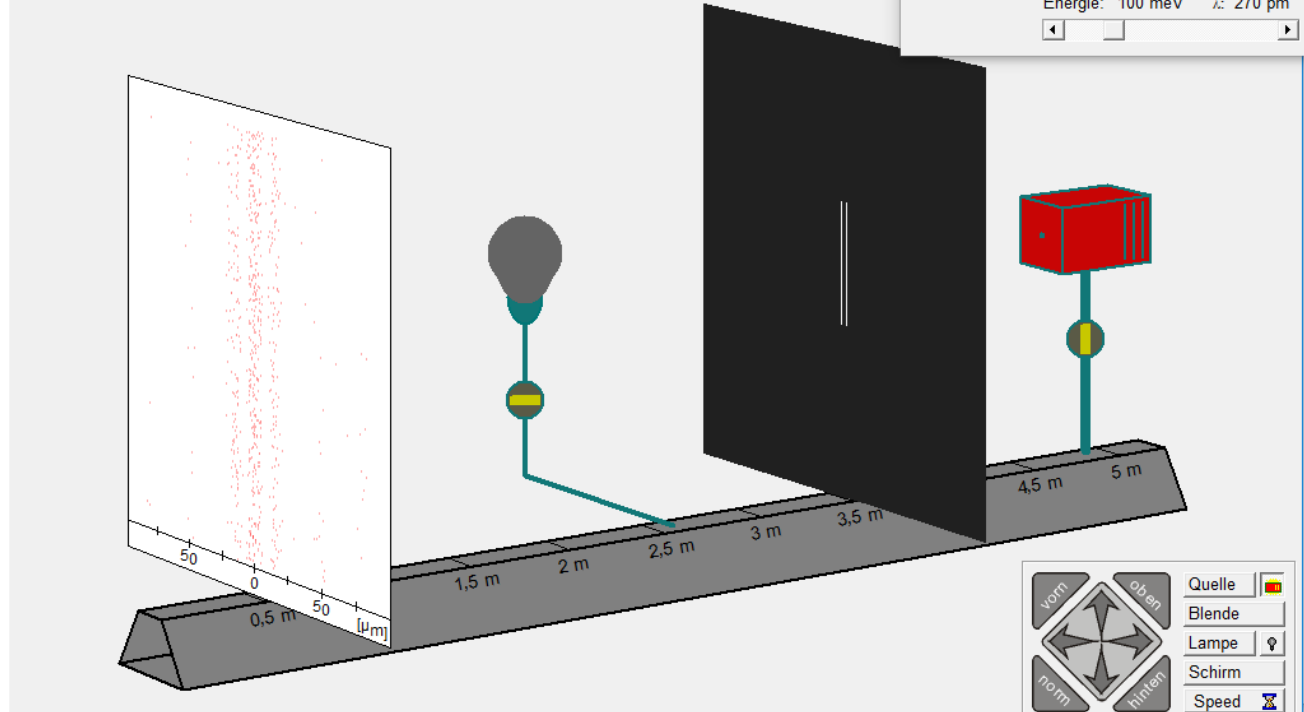
Quelle



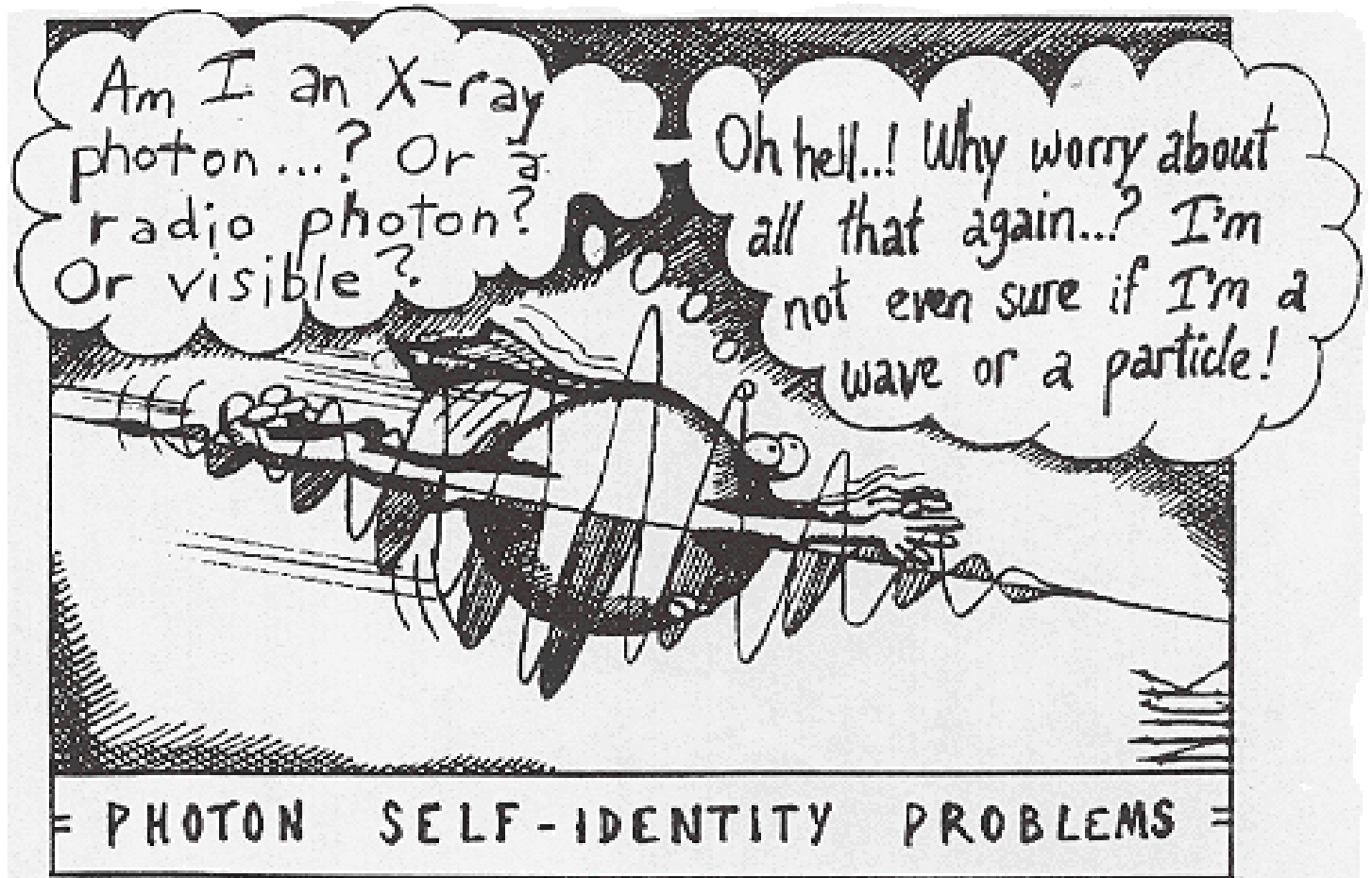
Teilchentyp:

- Kugeln
- Farbspray
- Photonen
- Elektronen
- Myonen
- Protonen
- Neutronen
- He-Atome
- Na-Moleküle
- Cs-Atome

Energie: 100 meV λ : 270 pm

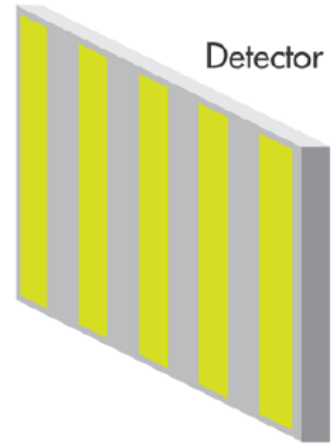
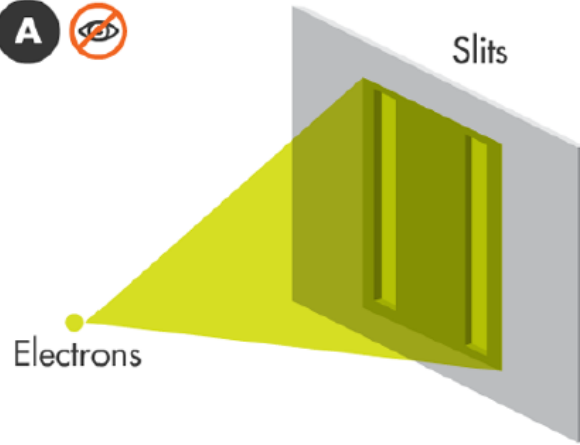


DoppelspaltversuchV4.0.exe

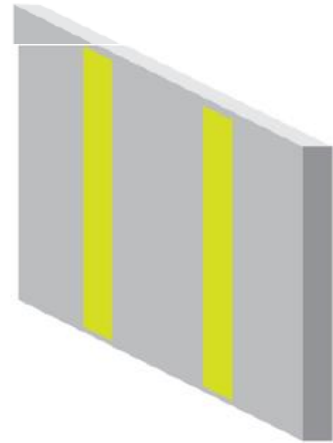
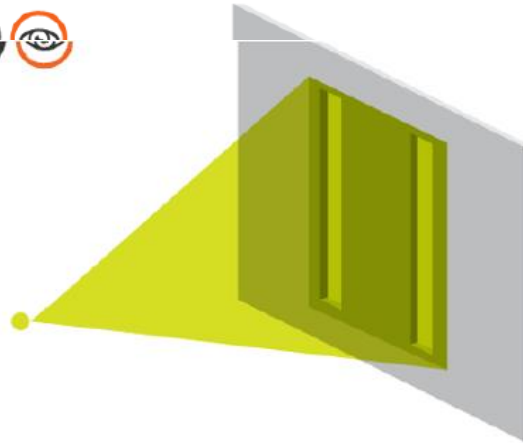


Dubbele spleet

A 

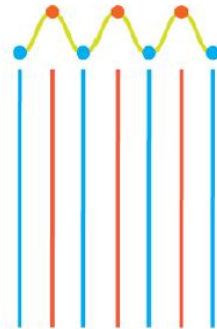


B 



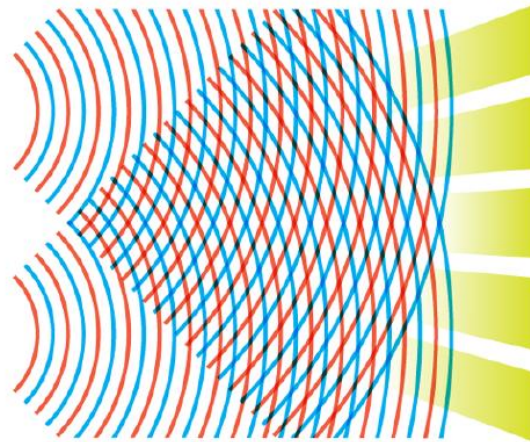
The Copenhagen Interpretation

So long as no observation is being made, electrons do not have definite positions. Each electron spreads out like a wave, passes through both slits simultaneously, and interferes with itself to form the bright and dark bands on the detector screen. Yet as soon as an observer checks to see which slit each electron is passing through, the observation instantaneously “collapses” each electron’s position to a point, thus ruining the interference pattern.



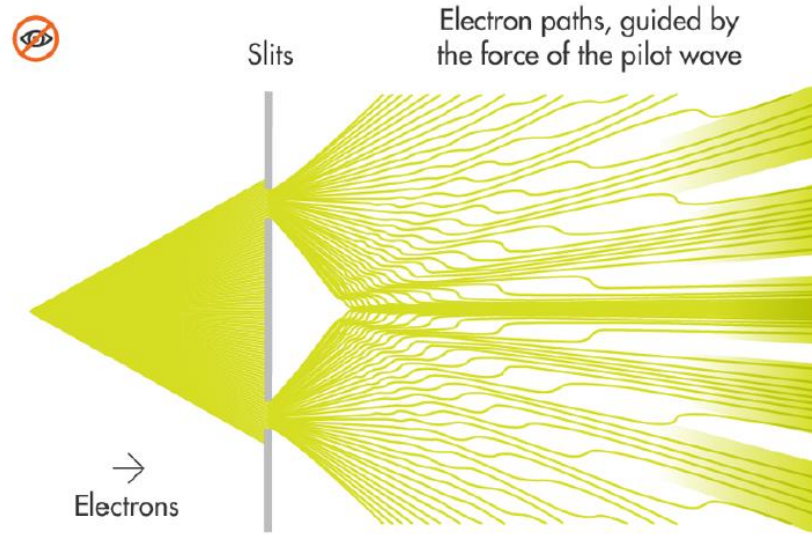
Slits

Where wave crests (red) meet troughs (blue), they cancel each other out



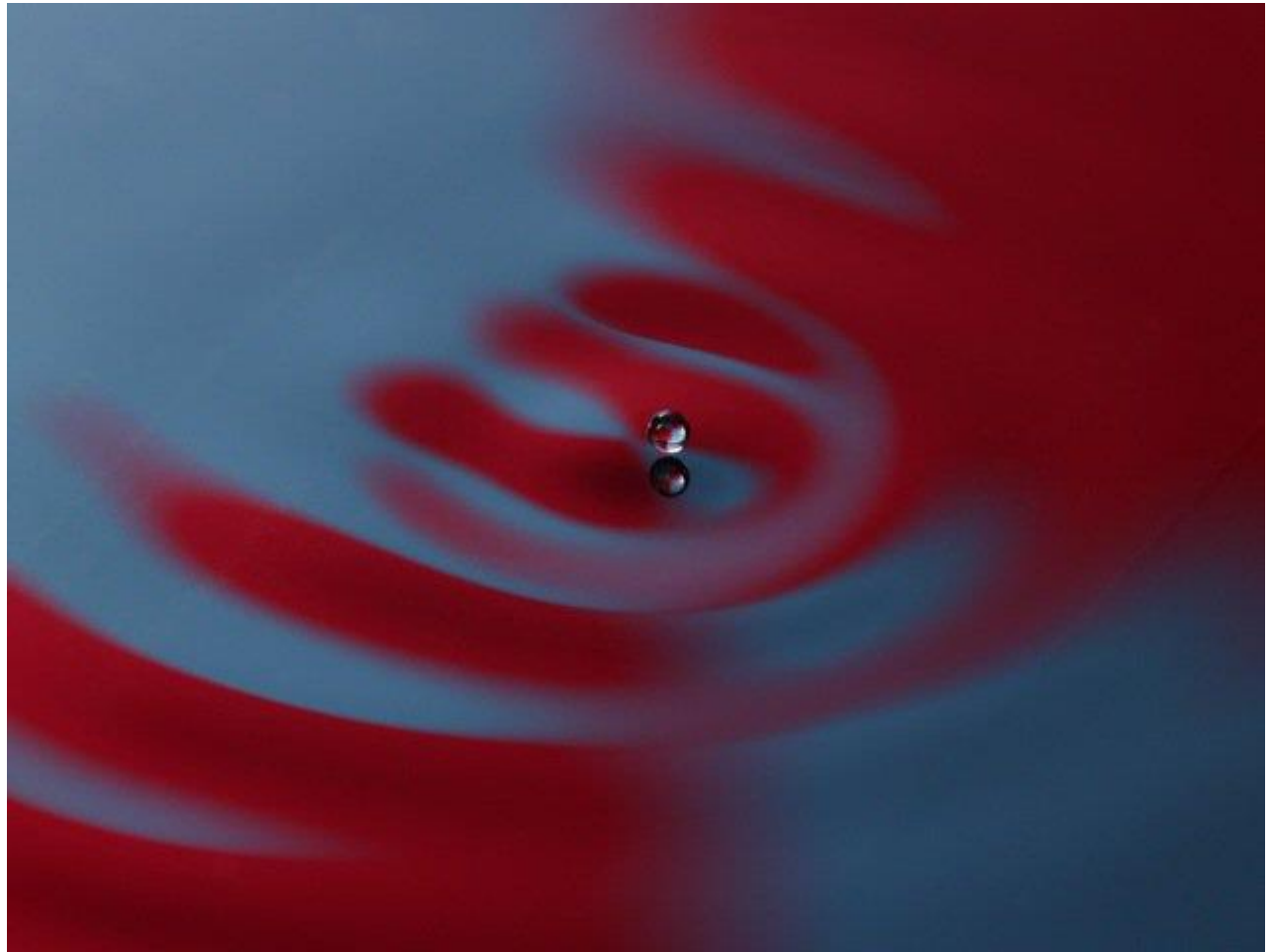
The Pilot-Wave Interpretation

In Bohmian mechanics, every electron always has a definite position, even if observers are ignorant of what it is. An electron is pushed around by a guiding “pilot wave” that influences the electron’s location. While each electron travels through one slit or the other, the pilot wave passes through both slits simultaneously. Interference in the pilot wave leads to the observed interference pattern. A measurement at the slits will collapse the pilot wave and reveal where the electron was all along.





Pilot wave



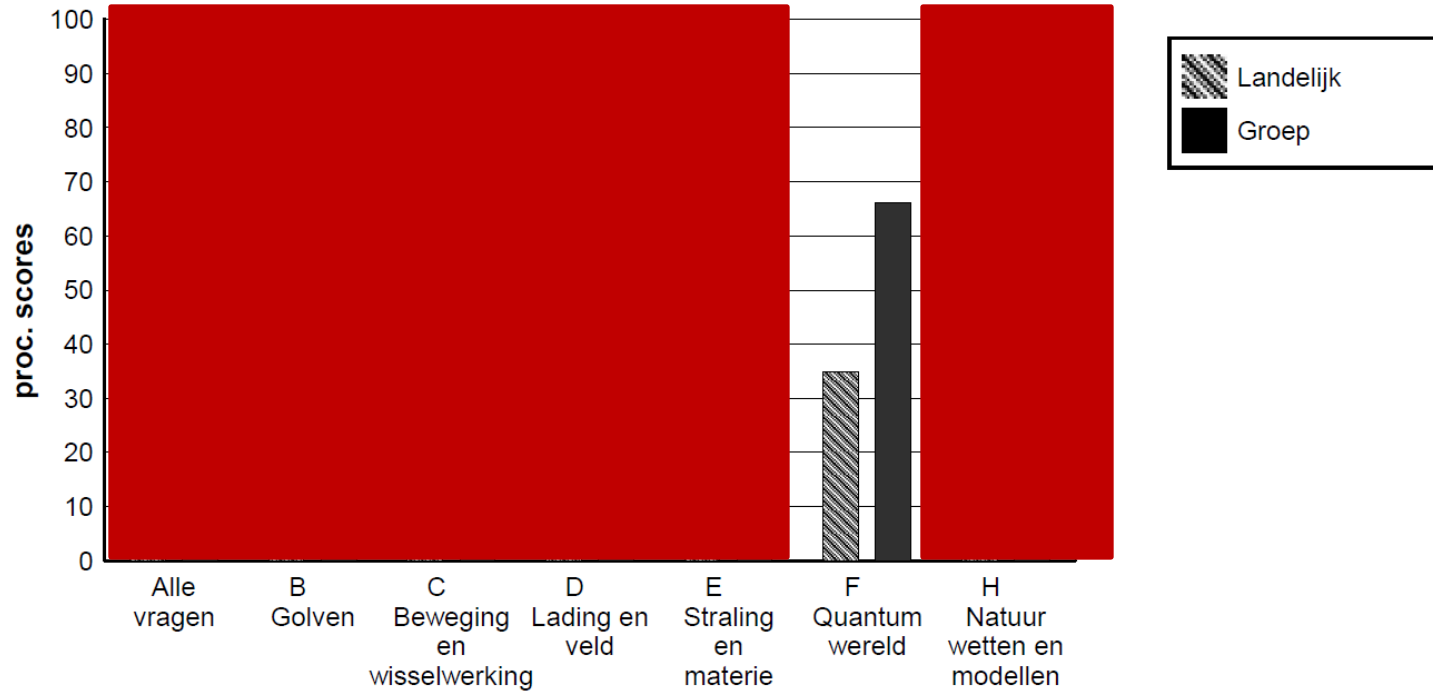


Pilot wave



wolf 2017

Domeinen uit het examenprogramma



Domeinen uit het examenprogramma

